



0040-4020(94)00403-X

Synthesis of Fused 1,2,4-Thiadiazoles from 5-Chloro-1,2,4-thiadiazol-3(2H)-ones

Gerrit L'abbé,* Johan Buelens, Wim Dehaen and Suzanne Toppet

Department of Chemistry, University of Leuven,
Celestijnenlaan 200F, 3001 Leuven (Heverlee), Belgium

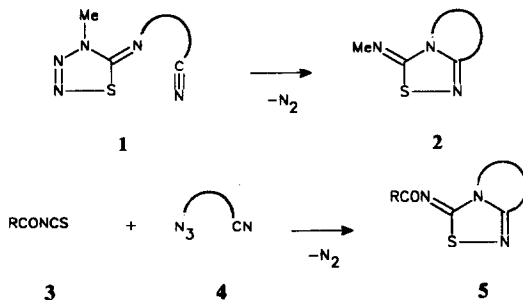
Janine Feneau-Dupont and Jean-Paul Declercq

Laboratoire de Chimie Physique et de Cristallographie,
Université Catholique de Louvain,
Place L. Pasteur 1, 1348 Louvain-La-Neuve, Belgique

Dedicated to Prof. L. Ghosez on the occasion of his 60th birthday.

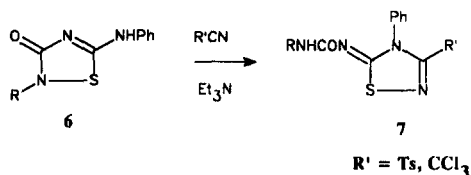
Abstract. Fused 1,2,4-thiadiazoles (**14-18** and **26-31**) are conveniently prepared by reacting the chlorothiadiazolone **12** with ω -aminonitriles and aminoazoles. During these reactions the original thiadiazole ring is opened and a new, fused thiadiazole ring is formed, probably via a hypervalent sulfur intermediate. The products derived from the aminoazoles were analyzed by X-ray crystallography and found to have structures different from those published earlier.⁷

In previous articles^{1,2} we have reported that fused 1,2,4-thiadiazoles are obtained by thermolysis of 5-(cyano tethered)imino-1,2,3,4-thiadiazolines (**1**→**2**) or by reacting acyl isothiocyanates with ω -azidonitriles (**3** + **4** → **5**). We have now investigated the reactions of 5-chloro-1,2,4-thiadiazol-3(2H)-ones with ω -aminonitriles and aminoazoles, leading also to fused thiadiazoles.

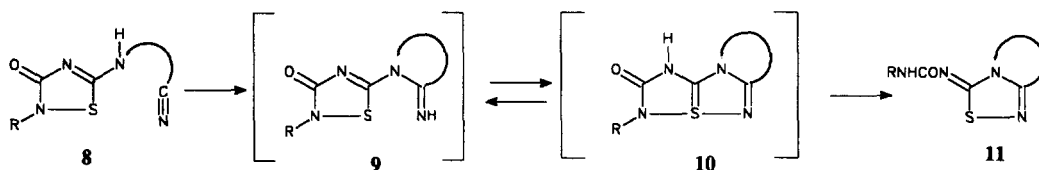


Our first approach is based on the knowledge that 5-amino-substituted 1,2,4-thiadiazol-3(2H)-ones **6** react with electrophilic nitriles to give rearranged thiadiazoles **7**.³ When these reactions are carried out intramolecularly, fused 1,2,4-thiadiazoles should result, and we surmised that the favourable entropy effect

might allow unactivated nitrile functions to react. According to our proposed Scheme 1, **9** and possibly also the hypervalent sulfur heterocycle **10**,⁴ are intermediates in the conversion of **8** to **11**. The results are discussed below.

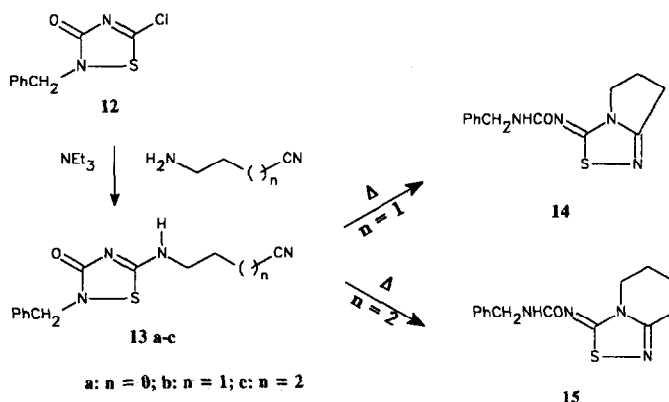


Scheme 1



The starting material **12**, used in this work, was prepared by chlorination of methoxymethyl isothiocyanate in the presence of benzyl isocyanate following the procedure of Keilen and Undheim.⁵ This compound reacted with 3-aminopropionitrile, 4-aminobutyronitrile and 5-aminovaleronitrile at room temperature to give the substitution products **13a-c**. Whereas **13a** remained unchanged when heated at 140°C, **13b** and **13c** rearranged at 60 °C in chloroform solution to give the fused heterocycles **14** and **15** respectively. They were characterized by IR, ¹H NMR, ¹³C NMR, mass spectra and microanalyses (see Experimental). In particular, the IR spectra were devoid of nitrile absorptions at about 2250 cm⁻¹ and the ¹H NMR spectra showed the presence of two doublets for the benzyl methylene protons due to restricted rotation around the amide side-chain.

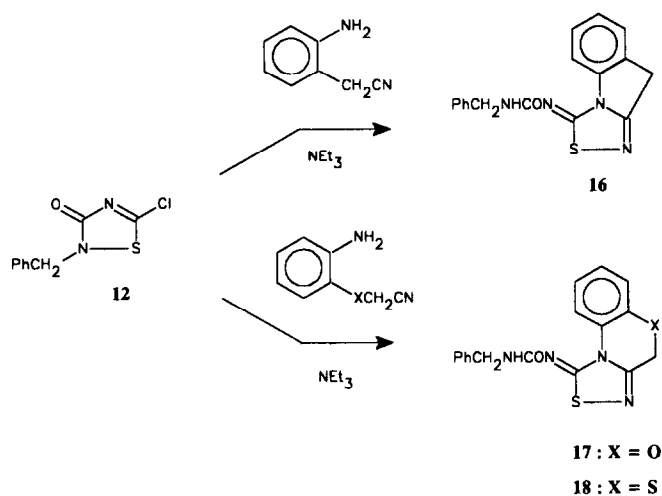
Scheme 2



The isomerizations of **13b,c** in DMSO- d_6 at 60 °C were followed kinetically by ^1H NMR spectroscopy and gave the following first-order rate constants: $k_1 = 3.25 \times 10^{-6} \text{ s}^{-1}$ for **13b**, and $k_1 = 2.14 \times 10^{-6} \text{ s}^{-1}$ for **13c**. Thus, the formation of a 5:5 fused heterocycle **14** is favoured over a 5:6 fused system **15**; and a 5:4 fused thiadiazole is not formed at all from **13a**. The same conclusion was reached for intramolecular cycloaddition-elimination reactions.¹

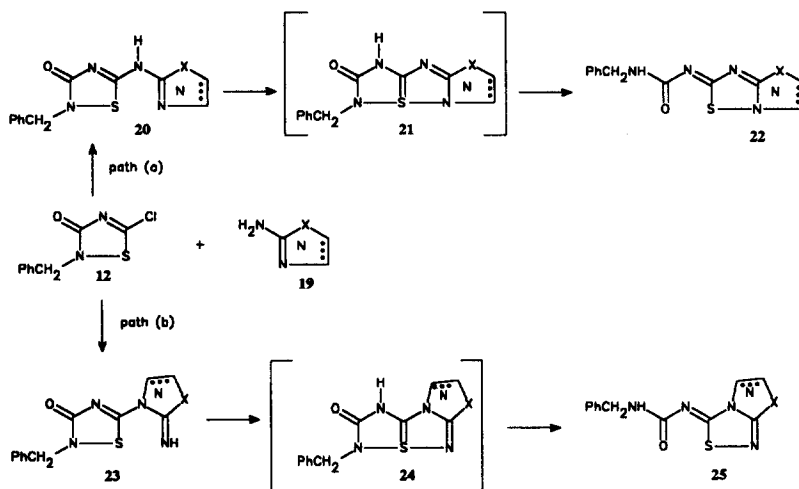
We also investigated the reactions of **12** with *o*-aminobenzyl cyanide, *o*-aminophenoxyacetonitrile and *o*-aminophenylthioacetonitrile, and obtained the rearranged products **16-18** directly at room temperature (Scheme 3). Here, the facility of the cycloaddition-ring opening reactions is caused by the mutual *ortho* disposition of the two interacting groups.

Scheme 3

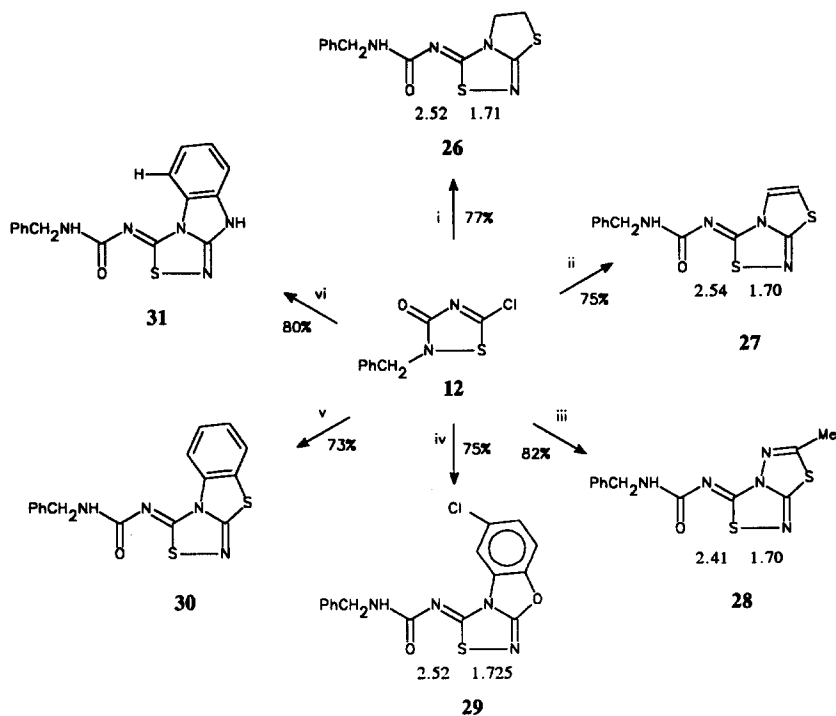


A second approach towards the synthesis of fused 1,2,4-thiadiazoles involves the reactions of **12** with aminoazoles **19** (Scheme 4). The latter may displace the chlorine atom of **12** either by the amine or the ring nitrogen atom, and evidence for both the possibilities is present in the literature.⁶ The two substitution products **20** and **23** are potential candidates for rearranging to fused thiadiazoles through the thiapentalenic intermediates **21** and **24**. In fact, **12** reacted with 2-amino-4,5-dihydrothiazole, 2-aminothiazole, 2-amino-5-methyl-1,3,4-thiadiazole, 2-amino-5-chlorobenzoxazole, 2-aminobenzothiazole and 2-aminobenzimidazole to give single products, whose spectral data were interpretable in terms of either structure **22** or **25**. Since no unequivocal ^{13}C NMR criterion could be found to distinguish between the two isomeric structures, we have subjected the products to a single crystal X-ray analysis. They all correspond to structure **25** (Scheme 5). In our preliminary communication on the subject,⁷ we had overlooked path (b) and attributed structure **22** to the products.

Scheme 4



Scheme 5



Reagents : i, 2-amino-4,5-dihydrothiazole; ii, 2-aminothiazole; iii, 2-amino-5-methyl-1,3,4-thiadiazole; iv, 2-amino-5-chlorobenzoxazole; v, 2-aminobenzothiazole; vi, 2-aminobenzimidazole

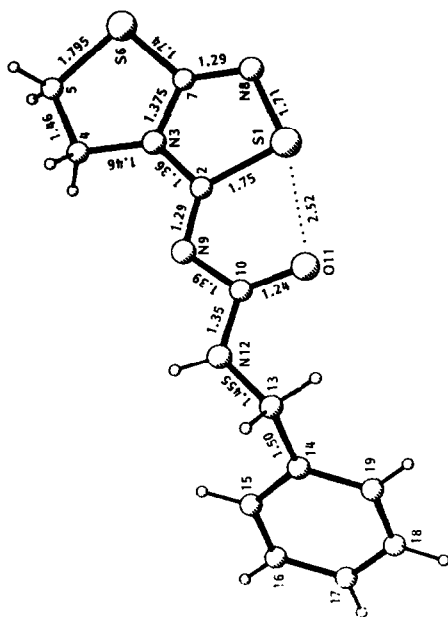


Fig. 1. Molecular structure of 26

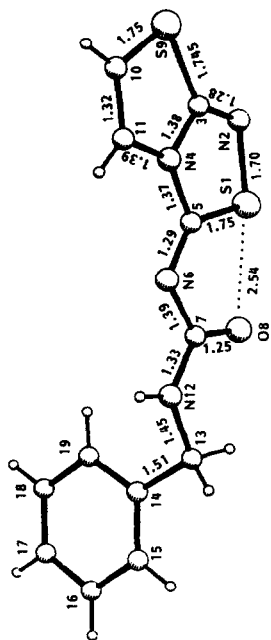


Fig. 2. Molecular structure of 27

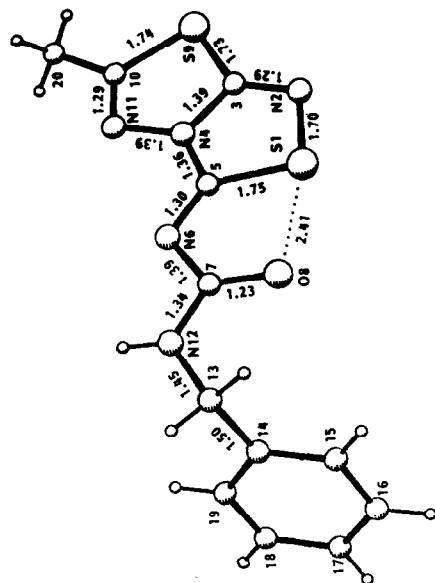


Fig. 3. Molecular structure of 28

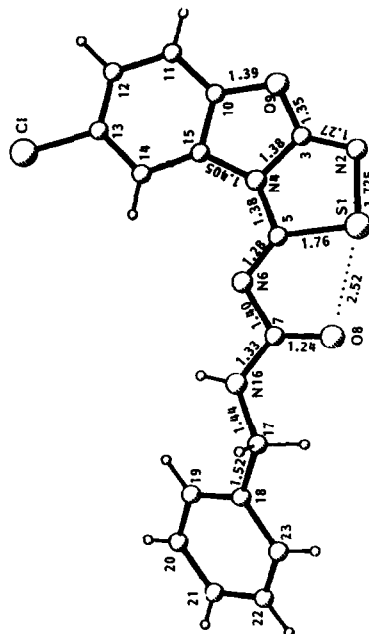
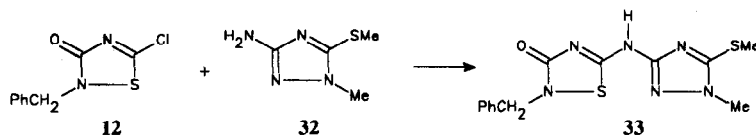


Fig. 4. Molecular structure of 29

The X-ray structure analyses revealed that the carbamoylimine substituent is nearly coplanar with the thiadiazole ring with a carbonyl-oxygen...sulfur distance ranging between 2.41 and 2.54 Å (Figs 1-4). This is significantly less than the sum of the corresponding van der Waals radii (3.2 Å) and slightly shorter than the Huggins constant energy distance of 2.58 Å,⁸ indicating little or no covalent bonding but rather a close interacting contact. The molecular structure of **30** is not included in this paper since the crystallographic data could not be refined due to much disorder in the crystal lattice; the available results, however, confirm the [1,2,4]thiadiazolo[3,4,b]benzothiazole skeleton of **30**.

No suitable crystals could be obtained from **31**, but its structure is suggested on the similarity of the NMR spectra with those of **29** and **30**. In particular, the benzo H-5 atom (indicated on the drawing) absorbs as two doublets, at δ 8.1 (major) and 7.95 (minor), separated by 0.15 ppm. This splitting is also found for the H-5 atoms of **29** and **30**, and is explained by the presence of two conformations of the side-chain amide function.

The only exception found to path (b) of Scheme 4 is the behaviour of **12** towards the aminotriazole **32** which yielded the substitution product **33**. This compound proved to be thermostable when heated in DMSO solution at 100 °C.



EXPERIMENTAL

Melting points were determined using a Reichert Thermovar apparatus. IR spectra were recorded on a Perkin Elmer 1720 FT spectrometer, NMR spectra on a Bruker WM-250 or AMX 400 spectrometer, and mass spectra (EI) on a Hewlett Packard 5989A or Kratos MS50 TC (for high resolution) instrument, both operating at 70 eV. X-Ray studies were carried out on a Huber 4-circle diffractometer with graphite-monochromatized MoK α ($\lambda = 0.71069$ Å) or CuK α radiation ($\lambda = 1.54178$ Å), and the structures were solved by direct methods¹² and refined by least-squares methods.

Synthesis of 13a-c: General procedure.

A solution of **12** (0.5 g, 2.2 mmol)⁵, ω -aminonitrile (1 equiv) and triethylamine (242 mg, 1.1 equiv) in dry chloroform (20 ml) was stirred at room temperature for 30 min. The reaction mixture was washed with water (3 x 10 ml), dried (MgSO₄) and evaporated, and the residue crystallized from ethanol.

2-Benzyl-5-(2-cyanoethylamino)-1,2,4-thiadiazol-3(2H)-one (13a) was obtained in 57 % yield, mp 155-156 °C (from EtOH); IR (KBr) 3250 (m, NH), 2250 (w, CN), 1642 cm⁻¹ (s, CO); ¹H NMR (250 MHz, DMSO-d₆) δ 2.8 (t, 2H, CH₂CN), 3.6 (br t, 2H, CH₂N), 4.7 (s, 2H, benzyl CH₂), 7.2-7.4 (m, 5H, Ph), 8.75 (br, 1H, NH); ¹³C NMR (DMSO-d₆) δ 17.2 and 38.9 (CH₂CH₂N), 46.3 (benzyl CH₂), 118.9 (CN), 127.7 (x 2), 128.6 and 137.3 (Ph C-atoms), 164.9 (C-3), 170.2 (C-5); mass spectrum, m/z (%) 260 (6, M⁺), 122 (30, CN(CH₂)₂N=C=NCO⁺), 106 (23), 91 (100, C₇H₇⁺). Anal. Calcd for C₁₂H₁₂N₄OS (mol wt 260.3): C, 55.37; H, 4.65. Found: C, 55.28; H, 4.61.

2-Benzyl-5-(3-cyanopropylamino)-1,2,4-thiadiazol-3(2H)-one (13b) was obtained in 55 % yield, mp 152-153 °C (from EtOH); IR (KBr) 3205 (m, NH), 2245 (m, CN), 1646 cm⁻¹ (s, CO); ¹H NMR (400 MHz, DMSO-d₆) δ 1.85 (quintet, 2H, C-CH₂-C),

2.55 (t, 2H, CH₂CN), 3.4 (br t, 2H, CH₂N), 4.7 (s, 2H, benzyl CH₂), 7.2-7.4 (m, 5H, Ph), 8.6 (br, 1H, NH); ¹³C NMR (DMSO-d₆) δ 13.8, 24.3 and 42.1 (CH₂CH₂CH₂N), 46.2 (benzyl CH₂), 120.0 (CN), 127.6 (x 2), 128.5 and 137.3 (Ph C-atoms), 165.2 (C-3), 169.6 (C-5); mass spectrum, m/z (%) 274 (60, M⁺), 168 (100, M⁺ - PhCH₂NH), 142 (17), 141 (12), 100 (19), 91 (29, C₇H₇⁺). Anal. Calcd for C₁₃H₁₄N₄OS (mol wt 274): C, 56.92; H, 5.14. Found: C, 56.88; H, 5.25.

2-Benzyl-5-(4-cyanobutylamino)-1,2,4-thiadiazol-3(2H)-one (13c) was obtained in 67 % yield, mp 134-134.5 °C (from EtOH); IR (KBr) 3232 (m, NH), 2247 (w, CN), 1644 cm⁻¹ (s, CO); ¹H NMR (400 MHz, DMSO-d₆) δ 1.5-1.65 (m, 4H, C-CH₂-CH₂-C), 2.5 (t, 2H, CH₂CN), 3.35 (br t, 2H, CH₂N), 4.7 (s, 2H, benzyl CH₂), 7.2-7.4 (m, 5H, Ph), 8.45 (br, 1H, NH); ¹³C NMR (DMSO-d₆) δ 15.7, 22.1, 27.6 and 42.5 (CH₂CH₂CH₂CH₂N), 46.3 (benzyl CH₂), 120.4 (CN), 127.6 (x 2), 128.5 and 137.3 (Ph C-atoms), 165.3 (C-3), 169.8 (C-5); mass spectrum, m/z 288 (30, M⁺), 182 (100, M⁺ - PhCH₂NH), 156 (19), 114 (20), 91 (48, C₇H₇⁺); M⁺ calcd: 288.1045; Found: 288.1037. Anal. Calcd for C₁₄H₁₆N₄OS (mol wt 288): C, 58.31; H, 5.59. Found: C, 58.24; H, 5.45.

3-(Benzylcarbamoylimino)-6,7-dihydro-3H,5H-pyrrolo[2,1-c][1,2,4]thiadiazole (14).

A solution of **13b** (0.5 g, 2.2 mmol) in dry chloroform (20 ml) was heated at 60 °C for 60 h. After removal of the solvent, the residue was crystallized from chloroform/diethyl ether to give **14** in 72 % yield (0.36 g), mp 169.5-171 °C (from EtOH); IR (KBr) 3400 (br, NH), 1606 cm⁻¹ (s); ¹H NMR (400 MHz, DMSO-d₆) δ 2.5 (quintet, 2H, H-6), 2.76 (t, 2H, H-7), 3.87 (t, 2H, H-5), 4.28 and 4.45 (two d in ratio 90:10, 2H, benzyl CH₂), 7.2-7.3 (m, 5H, Ph), 8.12 and 7.5 (two t in ratio 90:10, 1H, NH); ¹³C NMR (DMSO-d₆) δ 24.6 (C-6 and C-7) 43.5 (benzyl CH₂, ¹J_{CH} = 137 Hz), 44.3 (C-5), 126.6, 127.0, 128.2 and 140.2 (Ph C-atoms), 161.8 (C-7a), 163.3 (CO), 170.6 (C-3); mass spectrum, m/z (%) 274 (60, M⁺), 168 (100, M⁺ - PhCH₂NH), 142 (17, M⁺ - PhCH₂NCO), 100 (17), 91 (24, C₇H₇⁺); M⁺ calcd: 274.0888; Found: 274.0892. Anal. Calcd for C₁₃H₁₄N₄OS (mol wt 274): C, 56.92; H, 5.14. Found: C, 56.76; H, 5.09.

3-(Benzylcarbamoylimino)-5,6,7,8-tetrahydro-3H-[1,2,4]thiadiazolo[4,3-a]pyridine (15).

A solution of **13c** (0.5 g, 2.2 mmol) in dry chloroform (20 ml) was heated at 60 °C for 70 h. After removal of the solvent, the residue was crystallized from ethanol to give **15** in 55 % yield (0.35 g), mp 168-169.5 °C; IR (KBr) 3312 (m, NH) 1625 cm⁻¹ (s); ¹H NMR (400 MHz, DMSO-d₆) δ 1.83 (m, 2H, H-7), 1.93 (m, 2H, H-6), 2.75 (t, 2H, H-8), 3.85 (t, 2H, H-5), 4.30 and 4.48 (two d in ratio 90:10, ³J_{CH} = 6 Hz, 2H, benzyl CH₂), 7.2 - 7.35 (m, 5H, Ph), 8.05 and 7.5 (two t in ratio 90:10, 1H, NH); ¹³C NMR (DMSO-d₆) δ 19.3 and 21.4 (C-6 and C-7), 26.9 (C-8), 43.5 (benzyl CH₂), 45.7 (C-5), 126.5, 127.0, 128.1 and 140.2 (Ph C-atoms), 153.5 (C-8a), 163.1 (CO), 173.9 (C-3); mass spectrum, m/z (%) 288 (23, M⁺), 182 (100, M⁺ - PhCH₂NH), 156 (14), 114 (16), 91 (33, C₇H₇⁺). Anal. Calcd for C₁₄H₁₆N₄OS (mol wt 288): C, 58.31; H, 5.59. Found: C, 58.22; H, 5.53.

3-(Benzylcarbamoylimino)-3H,9H-[1,2,4]thiadiazolo[4,3-a]indole (16).

A solution of **12** (1.0 g, 4.4 mmol), o-aminobenzyl cyanide⁹ (581 mg, 4.4 mmol) and triethylamine (485 mg, 4.8 mmol) in dry ethanol (50 ml) was stirred at room temperature for 6 h. After removal of the solvent, the residue was washed successively with water (5 x 50 ml), ethanol (2 x 10 ml) and diethyl ether (2 x 10 ml), and dried in vacuo to give **16** in 48 % yield (660 mg), mp 203-204 °C (from EtOH); IR (KBr) 3309 (m, NH), 1621 cm⁻¹ (s); ¹H NMR (250 MHz, DMSO-d₆) δ 4.05 (s, 2H, H-9), 4.35 and 4.58 (two d in ratio 90:10, ³J_{CH} = 6 Hz, 2H, benzyl CH₂), 7.1-7.6 (m, 8 aromatic H), 8.20 and 8.0 (two d in ratio 90:10, 1H, H-5), 8.35 and 7.83 (two t in ratio 90:10, 1H, NH); ¹³C NMR (DMSO-d₆) δ 29.8 (C-9), 43.7 (benzyl CH₂), 113.6 (C-5), 126.0-128.2 (6 aromatic C-atoms), 132.0 (C-8a), 137.4 (C-4a), 139.8 (Ph Ci), 158.6 (C-9a), 163.0 (CO), 167.2 (C-3); mass spectrum, m/z (%) 322 (20, M⁺), 216 (79, M⁺ - PhCH₂NH), 190 (14, M⁺ - PhCH₂NCO), 162 (16), 132 (100, M⁺ -

PhCH₂NCO - NCS), 91 (57, C₇H₇⁺). Anal. Calcd for C₁₇H₁₄N₄OS (mol wt 322): C, 63.34; H, 4.38. Found: C, 63.09; H, 4.42.

3-(Benzylcarbamoylimino)-3*H*,10*H*-[1,2,4]thiadiazolo[3,4-*c*][1,4]benzoxazine (17).

A solution of 12 (1.0 g, 4.4 mmol), *o*-aminophenoxyacetonitrile¹⁰ (650 mg, 4.4 mmol) and triethylamine (485 mg, 4.8 mmol) in dry ethanol (50 ml) was stirred at room temperature for 3 h. The solvent was removed and the resulting brown oil was triturated with diethyl ether to give a solid residue. After washing successively with water (5 x 50 ml), ethanol (2 x 10 ml) and diethyl ether (2 x 10 ml), and drying in vacuo, 17 was obtained in 48 % yield (703 mg), mp 145.5-146.5 °C (from EtOH); IR (KBr) 3235 (m, NH), 1641 cm⁻¹ (s); ¹H NMR (250 MHz, DMSO-*d*₆) δ 4.35 and 4.50 (two d in ratio 90:10, ³J_{CH} = 6 Hz, 2H, benzyl CH₂), 5.15 and 5.25 (two s in ratio 90:10, 2H, H-10), 7.1-7.45 (m, 8 aromatic H), 8.55 and 8.0 (two t in ratio 90:10, 1H, NH), 9.1 and 8.6 (two d in ratio 90:10, 1H, H-5); ¹³C NMR (DMSO-*d*₆) δ 43.7 (benzyl CH₂), 64.7 (C-10), 117.3 (C-8), 120.3 (C-6), 122.7 (C-5), 125.6 (C-4a), 126.7, 126.8, 127.3 and 128.3 (aromatic C-atoms), 139.6 (Ph Ci), 146.2 (C-8a), 147.2 (C-10a), 162.7 (CO), 169.9 (C-3); mass spectrum, *m/z* (%) 338 (48, M⁺), 232 (100, M⁺ - PhCH₂NH), 206 (28), 178 (19), 148 (80, PhCH₂NHCON⁺), 106 (16, PhCH₂NH⁺), 91 (89, C₇H₇⁺). Anal. Calcd for C₁₇H₁₄N₄O₂S (mol wt 338): C, 60.34; H, 4.17. Found: C, 60.17; H, 4.11.

3-(Benzylcarbamoylimino)-3*H*,10*H*-[1,2,4]thiadiazolo[3,4-*c*][1,4]benzothiazine (18).

A solution of 12 (1.0 g, 4.4 mmol), *o*-aminophenylthioacetonitrile¹¹ (725 mg, 4.4 mmol) and triethylamine (485 mg, 4.8 mmol) in dry ethanol (50 ml) was stirred at room temperature for 3 h. The solvent was removed and the resulting brown oil was triturated with diethyl ether to give a solid residue. After washing successively with water (5 x 50 ml), ethanol (2 x 10 ml) and diethyl ether (2 x 10 ml), and drying in vacuo, 18 was obtained in 44 % yield (693 mg), mp 151-152 °C (from EtOH); IR (KBr) 3279 (m, NH), 1615 cm⁻¹ (s); ¹H NMR (250 MHz, DMSO-*d*₆) δ 4.07 (s, 2H, H-10), 4.3 and 4.4 (two d in ratio 83:17, ³J_{CH} = 6 Hz, 2H, benzyl CH₂), 7.2-7.5 (m, 7 aromatic H), 7.6 (d, 1H, H-8), 8.35 and 7.95 (two t in ratio 83:17, 1H, NH), 8.65 and 8.25 (two d in ratio 83:17, 1H, H-5); ¹³C NMR (DMSO-*d*₆) δ 28.3 (C-10), 43.6 (benzyl CH₂), 123.8 (C-5), 126.3-128.9 and 134.3 (8 aromatic C-atoms), 139.6 (Ph Ci), 150.5 (C-10a), 162.7 (CO), 171.2 (C-3); mass spectrum, *m/z* (%) 354 (35, M⁺), 321 (11), 248 (96, M⁺ - PhCH₂NH), 222 (26, M⁺ - PhCH₂NCO), 164 (62, M⁺ - PhCH₂NCO - NCS), 106 (14), 91 (100, C₇H₇⁺). Anal. Calcd for C₁₇H₁₄N₄OS₂ (mol wt 354): C, 57.61; H, 3.98. Found: C, 57.61; H, 4.04.

Synthesis of 26-31 and 33: General procedure.

A suspension of 12 (1.0 g, 4.4 mmol), aminoazole (1 equiv) and triethylamine (535 mg, 5.3 mmol) in dry ethanol (50 ml) was stirred at room temperature (1h for 26, 12 h for 28, 1 day for 31) or heated at 60 °C (2 h for 33, 12 h for 29 and 30 h for 27 and 30). After removal of the solvent, the residue was washed successively with water (5 x 50 ml), ethanol (2 x 10 ml) and diethyl ether (2 x 20 ml), and dried in vacuo.

3-(Benzylcarbamoylimino)-5,6-dihydro-3*H*-thiazolo[2,3-*c*][1,2,4]thiadiazole (26) was obtained from 2-amino-4,5-dihydrothiazole in 77 % yield, mp 163 °C (from EtOH); IR (KBr) 3310 (m, NH), 1635 cm⁻¹ (s); ¹H NMR (250 MHz, DMSO-*d*₆) δ 3.95 (t, 2H, H-6), 4.2 (t, 2H, H-5), 4.3 and 4.45 (two d in ratio 86:14, ³J_{CH} = 6 Hz, 2H, benzyl CH₂), 7.1-7.35 (m, 5H, Ph), 8.3 (t, 1H, NH); ¹³C NMR (DMSO-*d*₆) δ 33.5 (C-6), 43.5 (benzyl CH₂), 45.1 (C-5), 126.5, 127.0, 128.1 and 139.9 (Ph C-atoms), 156.8 (C-7a), 163.0 (CO), 170.2 (C-3); mass spectrum, *m/z* (%) 292 (54, M⁺), 186 (100, M⁺ - PhCH₂NH), 160 (25, PhCH₂NHCONC⁺), 159 (14, M⁺ - PhCH₂NCO), 106 (12, PhCH₂NH⁺), 91 (86, C₇H₇⁺). Anal. Calcd for C₁₂H₁₂N₄OS₂ (mol wt 292): C, 49.30; H, 4.14. Found: C, 49.14; H, 4.07. Crystal data: size 0.002 x 0.03 x 0.07 mm, orthorhombic space group Pbc_a with a = 7.789(4), b = 28.757(6) and c = 11.636(2) Å, Z = 8, V = 2606(2) Å³, D_x = 1.49

gcm^{-3} , $R = 0.054$ for 1723 observed reflections ($I > 2\sigma(I)$). Fig 1 shows a view of the molecule with an arbitrary non-systematic numbering scheme and selected bond lengths. The NCO group and thiadiazole ring are nearly coplanar, with a maximum deviation from the best plane through the eight atoms of 0.021 Å.

3-(Benzylcarbamoylimino)-3H-thiazolo[2,3-c][1,2,4]thiadiazole (27) was obtained from 2-aminothiazole in 75 % yield, mp 170-171 °C (from EtOH); IR (KBr) 3309 (m, NH); 1616 cm^{-1} (s); ^1H NMR (250 MHz, DMSO- d_6) δ 4.35 and 4.55 (two d in ratio 88:12, $^3J_{\text{CH}}$ = 6 Hz, 2H, CH_2), 7.28 and 7.52 (two d, $^3J_{\text{CH}}$ = 5 Hz, 2H, H-5 and H-6), 7.15-7.3 (m, 5H, Ph), 8.2 and 7.7 (two t in ratio 88:12, 1H, NH); ^{13}C NMR (DMSO- d_6) δ 43.7 (CH_2), 115.4 and 117.3 (C-5 and C-6), 126.6, 127.1, 128.1 and 139.7 (Ph C-atoms), 154.8 (C-7a, $^3J_{\text{CH}}$ = 6.5 and 7 Hz), 162.8 (CO), 168.3 (C-3); mass spectrum, m/z (%) 290 (25, M^+), 191 (17), 184 (32, M^+ - PhCH_2NH), 157 (19, M^+ - PhCH_2NCO), 130 (20, M^+ - $\text{PhCH}_2\text{NHCONC}$), 100 (76), 91 (100, C_7H_7^+). Anal. Calcd for $\text{C}_{12}\text{H}_{10}\text{N}_4\text{OS}_2$ (mol wt 290): C, 49.64; H, 3.47. Found: C, 49.45; H, 3.53. Crystal data: size 0.05 x 0.25 x 0.30 mm, orthorhombic space group $p2_12_12_1$ with $a = 8.049(2)$, $b = 11.731(3)$ and $c = 13.682(2)$ Å, $Z = 4$, $V = 1291.9(4)$ Å 3 , $D_x = 1.493$ gcm^{-3} , $R = 0.051$ for 2249 observed reflections ($I > 2\sigma(I)$). Fig 2 shows a view of the molecule with an arbitrary non-systematic numbering scheme and selected bond lengths. The NCO group and the thiadiazole ring are nearly coplanar, with a maximum deviation from the best plane through the eight atoms of 0.032 Å.

3-(Benzylcarbamoylimino)-6-methyl-3H-[1,3,4]thiadiazolo[2,3-c][1,2,4]thiadiazole (28) was obtained from 2-amino-5-methyl-1,3,4-thiadiazole in 82 % yield, mp 239-240 °C (from EtOH); IR (KBr) 3295 (m, NH), 1613 cm^{-1} (s); ^1H NMR (400 MHz, DMSO- d_6) δ 2.65 and 2.42 (two s in ratio 90:10, 3H, CH_3), 4.32 and 4.52 (two d in ratio 90:10, $^3J_{\text{CH}}$ = 6 Hz, 2H, CH_2), 7.2-7.4 (m, 5H, Ph), 8.52 and 7.85 (two t in ratio 90:10, 1H, NH); ^{13}C NMR (DMSO- d_6) δ 18.2 (CH_3), 43.7 (CH_2 , $^1J_{\text{CH}} = 140$ Hz), 126.7, 127.1, 128.3 and 139.8 (Ph C-atoms), 153.7 (C-7a), 162.5 (C-6), 162.9 (CO), 165.7 (C-3); mass spectrum, m/z (%) 305 (7, M^+), 199 (30, M^+ - PhCH_2NH), 191 (27), 173 (16, M^+ - PhCH_2NCO), 132 (13, PhCHNCO^+), 116 (18), 115 (21), 106 (17, PhCH_2NH^+), 91 (100, C_7H_7^+). Anal. Calcd for $\text{C}_{12}\text{H}_{11}\text{N}_5\text{OS}_2$ (mol wt 305): C, 47.20; H, 3.63. Found: C, 47.08, H, 3.53. Crystal data: size 0.20 x 0.18 x 0.03 mm, monoclinic space group C2/c with $a = 25.518(7)$, $b = 7.338(2)$, $c = 14.781(9)$ Å and $\beta = 105.67(7)^\circ$, $Z = 8$, $V = 2665(2)$ Å 3 , $D_x = 1.522$ gcm^{-3} , $R = 0.063$ for 1342 observed reflections ($I > 2\sigma(I)$). Fig 3 shows a view of the molecule with arbitrary non-systematic numbering scheme and selected bond lengths. The NCO group and the 1,2,4-thiadiazole ring are coplanar, with a maximum deviation from the best plane through the eight atoms of 0.013 Å.

3-(Benzylcarbamoylimino)-6-chloro-3H-[1,2,4]thiadiazolo[3,4-b]benzoxazole (29) was obtained from 2-amino-5-chlorobenzoxazole in 75 % yield, mp 181-183 °C; IR (KBr) 3248 (m, NH), 1667 and 1635 cm^{-1} (s); ^1H NMR (250 MHz, DMSO- d_6) δ 4.35 and 4.55 (two d in ratio 90:10, 2H, CH_2), 7.15-7.4 (m, 5H, Ph), 7.5 (dd, 1H, H-7), 7.65 (d, 1H, H-8), 7.85 and 7.70 (two d in ratio 90:10, 1H, H-5), 8.55 and 8.05 (two t in ratio 90:10, 1H, NH); ^{13}C NMR (DMSO- d_6) δ 43.7 (CH_2), 112.7 (C-5), 113.3 (C-8), 126.0-128.3 (6 aromatic C-atoms), 139.4 (Ph Cl), 149.2 (C-8a), 152.2 (C-9a), 162.8 (CO), 166.0 (C-3). mass spectrum, m/z (%) 358 (4, M^+), 252 (12, M^+ - PhCH_2NH), 168 (28), 148 (19), 91 (100, C_7H_7^+). Anal. Calcd for $\text{C}_{16}\text{H}_{11}\text{ClN}_4\text{O}_2\text{S}$ (mol wt 358): C, 53.56; H, 3.09. Found: C, 53.40; H, 3.10. Crystal data: size 0.12 x 0.12 x 0.35 mm, monoclinic space group $\text{P2}_1/a$ with $a = 8.294(2)$, $b = 12.2660(14)$, $c = 15.419(2)$ Å and $\beta = 101.61(2)^\circ$, $Z = 4$, $V = 1536.5(4)$ Å 3 , $D_x = 1.551$ gcm^{-3} , $R = 0.045$ for 1732 observed reflections ($I > 2\sigma(I)$). Fig 4 shows a view of the molecule with arbitrary non-systematic numbering scheme and selected bond lengths. The NCO group and the thiadiazole ring are almost coplanar, with a maximum deviation from the best plane through the eight atoms of 0.049 Å.

3-(Benzylcarbamoylimino)-3H-[1,2,4]thiadiazolo[3,4-b]benzothiazole (30) was obtained from 2-aminobenzothiazole in 73 % yield, mp 206-207 °C (from EtOH); IR (KBr) 3325 (m, NH), 1627 cm^{-1} (s); ^1H NMR (250 MHz, DMSO- d_6) δ 4.4 and 4.6 (two d in ratio 8:1, $^3J_{\text{CH}}$ = 6 Hz, 2H, CH_2), 7.1-7.6 (m, 7 aromatic H), 7.9 and 7.75 (two d in ratio 8:1, 1H, H-5), 8.55 and 8.0 (two t in ratio 8:1, 1H, NH), 8.73 and 8.50 (two d in ratio 8:1, 1H, H-8); ^{13}C NMR (DMSO- d_6) δ 43.7 (CH_2), 115.5,

124.0, 126.5-128.0, 131.5 and 139.4 (aromatic C-atoms), 152.9 (C-9a), 162.5 (CO), 168.4 (C-3); mass spectrum, *m/z* (%) 340 (43, M^+), 234 (50, M^+ - PhCH₂NH), 207 (11, M^+ - PhCH₂NCO), 180 (27, M^+ - PhCH₂NHCONC), 150 (86), 148 (11), 91 (100, C₇H₇⁺). Anal. Calcd for C₁₆H₁₂N₄OS₂ (mol wt 340): C, 56.45; H, 3.55. Found: C, 56.23; H, 3.56

3-(Benzylcarbamoylimino)-3*H*,9*H*-[1,2,4]thiadiazolo[4,3-*a*]benzimidazole (31) was obtained from 2-aminobenzimidazole in 80 % yield, mp 230-231 °C (from EtOH); IR (KBr) 3429 (m, NH), 3100 (br, NH), 1645 cm⁻¹ (s); ¹H NMR (250 MHz, DMSO-*d*₆) δ 4.35 and 4.6 (two d in ratio 90:10, ³J_{CH} = 6 Hz, 2H, CH₂), 7.1-7.4 (m, 8 aromatic H), 8.1 and 7.95 (two d in ratio 90:10, 1H, H-5), 8.25 and 7.75 (two t in ratio 90:10, 1H, NH), 11.9 (br, 1H, NH); ¹³C NMR (DMSO-*d*₆) δ 43.6 (CH₂), 110.8, 112.9, 120.7, 123.8, 125.9, 126.8, 127.2, 128.1, 136.5 and 139.9 (aromatic C-atoms), 148.1 (C-9a), 163.0 (CO), 167.2 (C-3); mass spectrum, *m/z* (%) 323 (7, M^+), 217 (23, M^+ - PhCH₂NH), 190 (10, M^+ - PhCH₂NCO), 163 (34, M^+ - PhCH₂NHCONC), 133 (92, PhCH₂NCO⁺), 105 (16, PhCH₂N⁺), 104 (21), 91 (100, C₇H₇⁺). Anal. Calcd for C₁₆H₁₃N₅OS (mol wt 323): C, 59.43; H, 4.05. Found: C, 59.21; H, 4.17.

2-Benzyl-5-(1-methyl-5-methylthio-1,2,4-triazol-3-yl)amino-1,2,4-thiadiazol-3(2*H*)-one (33) was obtained from 3-amino-1-methyl-5-methylthio-1,2,4-triazole in 66 % yield, mp 203-204 °C; IR (KBr) 3200-2600 (br, NH), 1691 (s, CO), 1616 cm⁻¹ (s); ¹H NMR (400 MHz, DMSO-*d*₆) δ 2.6 (s, 3H, MeS), 3.6 (s, 3H, CH₃), 4.7 (s, 2 H, CH₂), 7.3-7.4 (m, 5H, Ph), 12.2 (br, 1H, NH); ¹³C NMR (C₅D₅N at 90 °C) δ 15.8 (MeS), 32.3 (Me), 47.3 (CH₂), 128.2, 128.4, 129.2 and 137.9 (Ph C-atoms), 152.4 (triazole C-5), 159.8 (thiadiazole C-3), 161.2 (triazole C-3), 163.7 (thiadiazole C-5); mass spectrum, *m/z* (%) 334 (16, M^+), 196 (28), 106 (21, PhCH₂NH⁺), 91 (100, C₇H₇⁺). Anal. Calcd for C₁₃H₁₄N₆OS₂ (mol wt 334): C, 46.69; H, 4.22. Found: C, 46.44, H, 4.11.

REFERENCES AND NOTES

- L'abbé, G.; Leurs, S.; *J. Chem. Soc. Perkin Trans. 1*, **1992**, 181-182; L'abbé, G.; Leurs, S.; *Tetrahedron* **1992**, *48*, 7505-7518
- L'abbé, G.; Sannen, I.; Dehaen, W.; *J. Chem. Soc. Perkin Trans. 1*, **1993**, 27-29
- L'abbé, G.; Albrecht, E.; *J. Heterocycl. Chem.* **1992**, *29*, 451-454.
- Akiba, K.; Arai, S.; Tsuchiya, T.; Yamamoto, Y.; Iwasaki, F.; *Angew. Chem.* **1979**, *91*, 176-177; *Angew. Chem. Int. Ed. Engl.* **1979**, *18*, 166-167; Akiba, K.; Kobayashi, T.; Arai, S.; *J. Am. Chem. Soc.* **1979**, *101*, 5857; L'abbé, G.; Weyns, N.; Sannen, I.; Delbeke, P.; Toppet, S.; *J. Heterocycl. Chem.* **1991**, *28*, 405-409; L'abbé, G.; Weyns, N.; *Bull. Soc. Chim. Belg.*; **1991**, *100*, 185-186; L'abbé, G.; Vandendriessche, A.; Sannen I.; *J. Org. Chem.* **1991**, *56*, 3268-3270. Lai, L.; Ngoi, T.; Reid, D.H.; Nicol, R.H.; Rhodes, J.B.; *J. Chem. Soc. Perkin Trans. 1*, **1993**, 1753-1759.
- Keilen, G.; Undheim, K.; *Acta Chem. Scand. Ser. B* **1988**, *42*, 362-366.
Note: The ring carbon absorptions of compound 12 in the ¹³C NMR spectrum (CDCl₃) were erroneously assigned by these authors; they should read C-3 at δ 162.8 (³J_{CH} = 2.5 Hz) and C-5 at δ 166.8
- Mitchell, J.A.; Reid, D.H.; *J. Chem. Soc. Perkin Trans 1*, **1982**, 499-507.
- L'abbé, G.; Buelens, J.; Dehaen, W.; *J. Chem. Soc. Perkin Trans. 1*, **1993**, 1825-1826.
- Huggins, M.L.; *J. Am. Chem. Soc.* **1953**, *75*, 4126-4133.
- Rousseau, V.; Lindwall, H.G.; *J. Am. Chem. Soc.* **1950**, *72*, 3047-3051.
- Mazharuddin, M.; Thyagarajan, G.; *Chem. Ind. (London)* **1971**, 178.
- Garanti, L.; Scandroglio, A.; Zecchi, G.; *J. Heterocycl. Chem.* **1976**, *13*, 1339-1341.
- Sheldrick, G.M.; SHELXS 86. *Acta Cryst.*, **1990**, *A46*, 467-473.

(Received in UK 7 February 1994; accepted 11 May 1994)